Benzastatins H and I, New Benzastatin Derivatives with Neuronal Cell Protecting Activity from *Streptomyces nitrosporeus*

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L-Glutamate, a major neurotransmitter in the central nervous system, has been known to be extensively released during brain ischemia and induces subsequent neuronal cell death^{1,2)}. Recent studies indicate that oxygen radicals are produced through a variety of intracellular cascades in such events²⁾. It was also reported that blockage of glutamate toxicity by free radical scavengers was effective to ameliorate brain ischemia injury^{3,4)}. Recently, some glutamate toxicity inhibitors of microbial origin such as carquinostatin A⁵), lavanduquinocin⁶), and aestivophoenins A and B^{7} have been reported. In the course of our screening for free radical scavengers or inhibitors of glutamate toxicity using the neuronal hybridoma N18-RE-105 cells to prevent the brain ischemia injury, we $A \sim G^{8 \sim 10}$ isolated benzastatins previously and phenazostatins $A \sim C^{11 \sim 13}$. Further investigation on polar metabolites of Streptomyces nitrosporeus 30643 which is the producer of benzastatins A~G has resulted in isolation of two hydroxylated derivatives of benzastatin B (3), benzastatins H (1) and I (2) (Fig. 1). We report here the isolation, physico-chemical properties, structure determination, and biological activities of 1 and 2.

The EtOAc extract from the broth filtrate (26 liters) of *S. nitrosporeus* 30643 was subjected to SiO₂ (Merck art No. 7734.9025) column chromatography followed by elution with hexane-EtOAc (1:2) containing 0.5% of conc. NH₄OH. The active fractions were pooled and concentrated *in vacuo* to give an oily residue. The residue was applied again to a SiO₂ column and then eluted with hexane-EtOAc (1:6) containing 0.5% of conc. NH₄OH. Active fraction dissolved in MeOH was further purified by reverse phase HPLC column (22.6×300 mm, Phenomenex C₁₈, USA) chromatography with a photodiode array detector. The column was eluted with CH₃CN-H₂O (29:71) at a

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flow rate of 8 ml/minute to afford two structurally related new compounds, benzastatin H (1, 1.4 mg) with a retention time of 24 minutes and benzastatin I (2, 1.1 mg) at 25 minutes.

The physico-chemical properties of 1 and 2 are summarized in Table 1. They are soluble in methanol and dimethlylsulfoxide, slightly soluble in acetonitrile, and insoluble in water, acetone, ethyl acetate, chloroform, and *n*-hexane. After TLC on silica gel 60 F_{254} (Merck) with chloroform - methanol (10:1) containing 0.5% of conc. NH₄OH, 1 and 2 showed the same Rf value of 0.27 whereas 3 had an Rf value of 0.4. The UV absorption spectra of 1 and 2 showed the same absorption maxima at 206 and 286 nm which was very similar to that of 3. The IR spectra of 1 and 2 revealed the characteristic absorption bands of an amide carbonyl group (1651 cm⁻¹).

The molecular formula of 1 was determined to be $C_{18}H_{26}N_2O_2$ on the basis of high resolution EI-MS [M⁺, m/z 302.1997 (+0.3 mmu error) in combination with ¹H and ¹³C NMR data. Together with UV and IR spectral data, the ¹H and ¹³C NMR spectra of **1** were similar to those of **3** (Table 2). Comparison of ¹H and ¹³C NMR data with HMQC data between 1 and 3 revealed that a methylene signal [$\delta_{\rm H}$ 4.12 (2H, s, H₂-17) and $\delta_{\rm C}$ 61.5 (C-17)] newly appeared instead of the allylic methyl of **3**. The 1 H and 13 C chemical shifts of the methylene signal suggest that one of four allylic methyls of 3 was hydroxylated in 1, which was supported by HREI-MS data showing the molecular formula of 1 with one more oxygen than that of 3. The position of the hydroxylated methylene was determined by HMBC and NOESY experiments (Fig. 2). Long range couplings were observed from the hydroxylated methylene





	1	2
Appearance	white powder	white powder
EI-MS (m/z)	302 (M) ⁺	302 (M) ⁺
HREI-MS (m/z)		
found	302.1997	302.1991
calcd.	302.1994	302.1994
Molecular formula	C ₁₈ H ₂₆ N ₂ O ₂	C ₁₈ H ₂₆ N ₂ O ₂
UV λ_{max} nm (log ϵ)	206(4.27), 286 (4.11)	206(4.32), 286 (4.14)
IR (KBr)v cm ⁻¹	3385, 2926, 1651,	3356, 2925, 1651,
	1601, 1383	1601, 1383

Table 1. Physico-chemical properties of 1 and 2.

Table 2. ¹H and ¹³C NMR spectral data for 1, 2 and 3.

Position	1 (CD ₃ OD)		2 (CD ₃ OD)		3 (CDCl ₃) ⁹⁾	
	δ _H	δ _C	$\delta_{\rm H}$	δ _C	$\delta_{\rm H}$	δ _C
1		122.5		122.4		122.8
2	7.62 (1H, d, 2.1) ^a	130.0	7.57 (1H, d, 2.0)	129.8	7.56 (1H, d, 1.8)	129.5
3		125.1		125.4		125.1
4		150.5		150.7		148.6
5	6.73 (1H, d, 8.3)	114.6	6.68 (1H, d, 8.3)	114.8	6.65 (1H, d, 8.2)	114.6
6	7.56 (1H, dd, 8.3, 2	.1)127.6	7.52 (1H, dd, 8.3, 2	.0)127.6	7.52 (1H, dd, 8.2, 1	.8)126.9
7		172.5		172.6		169.4
8	3.28 (2H, d, 7.2)	30.6	3.23 (2H, d, 7.2)	30.5	3.25 (2H, d, 6.5)	30.9
9	5.36 (1H, t, 7.2)	121.8	5.31 (1H, t, 7.2)	122.6	5.20 (1H, t, 6.5)	120.7
10		138.4		138.0		138.5
11	2.17 (2H, m)	39.1	2.15 (2H, m)	39.7	2.05 (2H, m)	38.1
12	2.33 (2H, m)	30.2	2.25 (2H, m)	33.5	2.15 (2H, m)	33.4
13		132.7		133.5		127.3
14		130.2		128.8		124.3
15	1.77 (3H, s)	19.9	1.70 (3H, s)	16.2	1.63 (3H, s)	20.6
16	1.84 (3H, s)	16.0	1.77 (3H, s)	15.9	1.77 (3H, s)	16.4
17	4.12 (2H, s)	61.5	1.69 (3H, s)	18.6	1.62 (3H, s)	18.3
18	1.74 (3H, s)	20.5	4.03 (2H, s)	61.8	1.64 (3H, s)	20.1

All spectra of 1 and 2 were recorded at 300 MHz for ¹H and 125 MHz for ¹³C. ^aProton resonance multiplicity and coupling constant (J = Hz) are in parenthesis. The assignments were aided by NOESY. HMOC, and HMBC

The assignments were aided by NOESY, HMQC, and HMBC.

protons (H₂-17) to one methylene carbon at δ 30.2 (C-12) and two *sp*² quaternary carbons at δ 132.7 (C-13) and δ 130.2 (C-14). In addition, NOEs were observed from the hydroxylated methylene protons (H₂-17) to H₂-11, H₂-12, and H₃-15. These spectral data indicate that the hydroxylated methylene should be attached at C-13. The remaining structure of **1** was also confirmed by the HMBC spectral data as shown in Fig. 2. Thus, **1** was determined to be a derivative hydroxylated at C-17 of **3**.

The molecular formula of **2** was determined to be $C_{18}H_{26}N_2O_2$ on the basis of high resolution EI-MS [M⁺, *m/z* 302.1991 (-0.3 mmu error)] in combination with ¹H and ¹³C NMR data. The molecular formula of **2** was the same as that of **1**. Together with UV and IR spectral data, the ¹H and ¹³C NMR spectra of **2** were very similar to those of **1** (Table 2). The only difference in ¹H and ¹³C NMR data with

Fig. 2. HMBC and NOE data of 1 and 2.



HMQC data was that ¹H and ¹³C chemical shifts of three allylic methyls in 2 were a little bit different from those of 1, suggesting that another allylic methyl may be hydroxylated in 2. The position of the hydroxylated methylene was determined by HMBC and NOESY experiments. Long range couplings were observed from the hydroxylated methylene protons (H₂-18) to two sp^2 quaternary carbons at δ 133.5 (C-13) and δ 128.8 (C-14) and one allylic methyl carbon at 16.2 (C-15), not the methylene carbon at 30.2 (C-12). Instead, one allylic methyl at δ 18.6 (C-17) was long range coupled to C-12, C-13, and C-14. In addition, NOEs were observed from H₃-17 to H₂-11 and H₂-12, and from the hydroxylated methylene protons (H₂-18) to H₂-12 and H₃-15. These spectral data indicate that 2 is a derivative hydroxylated at C-18 of 3. The remaining structure of 2 was also confirmed by the HMBC spectral data in Fig. 2.

The protective effect of 1 and 2 on glutamate toxicity in neuronal N18-RE-105 cells^{14,15)} was examined. 1 and 2 protected the cells from glutamate toxicity in a dose dependant fashion with EC₅₀ values of 30.3 and 21.6 μ M, respectively. The inhibition activity of 1 and 2 was similar to that of 3. Idebenone¹⁶⁾, a known brain protective agent with free radical scavenging activity, which was used as a positive control, showed EC₅₀ value of 0.7 μ M. 1 and 2 did not show cytotoxicity at 200 μ M while idebenone exhibited a strong cytotoxicity with an IC_{50} value of 4.0 μ M in this assay system.

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